

## Documents

Cherifi, K.<sup>a, b</sup>, Cheknane, A.<sup>b</sup>, Benghia, A.<sup>c</sup>, Hilal, H.S.<sup>d</sup>, Rahmoun, K.<sup>a</sup>, Benyoucef, B.<sup>a</sup>, Goumri-Said, S.<sup>e</sup>

**Exploring N3 ruthenium dye adsorption onto ZnTiO<sub>3</sub> (101) and (110) surfaces for dye sensitized solar cell applications: Full computational study**

(2019) *Materials Today Energy*, 13, pp. 109-118.

DOI: 10.1016/j.mtener.2019.04.013

<sup>a</sup> Unité de Recherche Matériaux et énergies Renouvelables-URMER, Université Abou Bekr Belkaid de Tlemcen, BP 119, Tlemcen, 13000, Algeria

<sup>b</sup> Laboratoire des Semi-conducteurs et Matériaux Fonctionnels, Université Amar Telidji de Laghouat, Bd des Martyrs BP37G, Laghouat, 03000, Algeria

<sup>c</sup> Laboratoire de Physique des Matériaux, Université Amar Telidji de Laghouat, Bd des Martyrs BP37G, Laghouat, 03000, Algeria

<sup>d</sup> SSERL, Chemistry Department, An-Najah National University, Nablus, Palestine

<sup>e</sup> College of Science, Physics Department, Alfaisal University, P.O. Box 50927, Riyadh, 11533, Saudi Arabia

**Abstract**

TiO<sub>2</sub> and ZnO binary oxide films are extensively investigated as photo-anodes in dye-sensitized solar cells (DSSCs), due to their large surface area, high electron mobility and chemical stability features. ZnTiO<sub>3</sub> perovskite is investigated as a photoanode, with N3 ruthenium dye, to design a new DSSC. First principal calculations based on DFT methods have been carried out to study the adsorption process of the ruthenium dye onto ZnTiO<sub>3</sub> (101) and (110) surfaces. The energy levels of N3 dye are studied and compared to available experimental data. The calculated energy band gap for ZnTiO<sub>3</sub> is 3.10 eV, which is in good agreement with experimental data (3.18 eV). The N3 dye is chemisorbed on the perovskite surface, via its carboxylic groups (COOH) which link to the (101) surface and two monodentate (ME) ester modes in (110) with bidentate bridging (BB) adsorption mode. The calculated adsorption energy (-241.50 kcal/mol) for the resulting N3@ZnTiO<sub>3</sub> complex, is much higher than the values for N3@ZnO (-89.58 kcal/mol) and N3@TiO<sub>2</sub> (-146.90 kcal/mol). The high adsorption energy of N3 dye, onto the perovskite surfaces, shows the potential value to use perovskites in future DSSCs. © 2019 Elsevier Ltd

**Author Keywords**

Adsorption energy; DFT calculation; DSSC; N3@ZnTiO<sub>3</sub> complex; Perovskite

**Index Keywords**

Adsorption, Chemical stability, Design for testability, Energy gap, II-VI semiconductors, Oxide films, Perovskite, Ruthenium, Solar cells, Titanium dioxide, Zinc oxide; Adsorption energies, Adsorption process, Computational studies, DFT calculation, DSSC, First-principal calculations, High electron mobility, N3@ZnTiO<sub>3</sub> complex; Dye-sensitized solar cells

**References**

- Jiao, Y., Zhang, F., Meng, S.  
**Dye sensitized solar cells principle and new Desi**  
(1991) *Int. J. Mol. Sci.*, 11 (3), pp. 5-37.
- O'Regan, B., Gratzel, M.  
**A low-cost, high-efficiency solar-cell based on dye-sensitized colloidal TiO<sub>2</sub> films**  
(1991) *Nature*, 353 (6346), pp. 737-740.
- Gong, J., Sumathy, K., Qiao, Q., Zhou, Z.  
**Review on dye-sensitized solar cells (DSSCs): advanced techniques and research trends**  
(2017) *Renew. Sustain. Energy Rev.*, 68 (September 2016), pp. 234-246.

- Jasim, K.E.  
**Dye sensitised solar cells-working principles, challenges and opportunities**  
(2007) *Dye Sensitized Sol. Cells – Work. Princ. Challenges*, pp. 171-204.
- Mishra, A., Fischer, M.K.R., Bäuerle, P.  
**Metal-free organic dyes for dye-sensitized solar cells: from structure: property relationships to design rules**  
(2009) *Angew. Chem. Int. Ed.*, 48 (14), pp. 2474-2499.
- Haque, M.A., Sheikh, A.D., Guan, X., Wu, T.  
**Metal oxides as efficient charge transporters in perovskite solar cells**  
(2017) *Adv. Energy Mater.*, 7 (20), pp. 1-23.
- Ke, W.  
**Low temperature solution-processed tin oxide as an alternative electron transporting layer for efficient perovskite solar cells**  
(2015) *J. Am. Chem. Soc.*, 137 (21), pp. 6730-6733.
- Han, G.S.  
**Retarding charge recombination in perovskite solar cells using ultrathin MgO-coated TiO<sub>2</sub> nanoparticulate films**  
(2015) *J. Mater. Chem. A*, 3 (17), pp. 9160-9164.
- Correa Baena, J.P.  
**Highly efficient planar perovskite solar cells through band alignment engineering**  
(2015) *Energy Environ. Sci.*, 8 (10), pp. 2928-2934.
- Mohamad, M., Ahmed, R., Kanoun, A.A., Shaari, A., Goumri-Said, S.  
**Designing a molecular device for organic solar cell applications based on Vinazene: IV characterization and efficiency predictions**  
(2016) *Sol. Energy*, 140, pp. 124-129.
- Manseki, K., Ikeya, T., Tamura, A., Ban, T., Sugiura, T., Yoshida, T.  
**Mg-doped TiO<sub>2</sub> nanorods improving open-circuit voltages of ammonium lead halide perovskite solar cells**  
(2014) *RSC Adv.*, 4 (19), pp. 9652-9655.
- Kanoun, A.-A., Kanoun, M.B., Merad, A.E., Goumri-Said, S.  
**Toward development of high-performance perovskite solar cells based on CH<sub>3</sub>NH<sub>3</sub>GeI<sub>3</sub> using computational approach**  
(2019) *Sol. Energy*, 182, pp. 237-244.
- Kojima, A., Teshima, K., Shirai, Y., Miyasaka, T.  
**Organometal halide perovskites as visible- light sensitizers for photovoltaic cells**  
(2009) *J. Am. Chem. Soc.*, 131 (October), pp. 6050-6051.
- Ilyassov, B.R., Ibraev, N.K., Abzhanova, D.B.  
**Effect of morphology of ZnO nanowire arrays on photovoltaic and electron transport properties of DSSC**  
(2015) *IOP Conf. Ser. Mater. Sci. Eng.*, 81 (1).
- Xu, F., Sun, L.  
**Solution-derived ZnO nanostructures for photoanodes of dye-sensitized solar cells**  
(2011) *Energy Environ. Sci.*, 4 (3), pp. 818-841.

- OKAMOTO, Y., SUZUKI, Y.  
**Perovskite-type SrTiO<sub>3</sub>, CaTiO<sub>3</sub> and BaTiO<sub>3</sub> porous film electrodes for dye-sensitized solar cells**  
(2014) *J. Ceram. Soc. Jpn.*, 122 (1428), pp. 728-731.
- Yu, J., Li, D., Zhu, L., Xu, X.  
**Application of ZnTiO<sub>3</sub> in quantum-dot-sensitized solar cells and numerical simulations using first-principles theory**  
(2016) *J. Alloy. Compd.*, 681, pp. 88-95.
- Lu, Y., Choi, D., Nelson, J., Yang, O.-B., Parkinson, B.A.  
**Adsorption, desorption, and sensitization of low-index anatase and rutile surfaces by the ruthenium complex dye N3**  
(2006) *J. Electrochem. Soc.*, 153 (8), p. E131.
- Pastore, M., De Angelis, F.  
**Computational modelling of TiO<sub>2</sub> surfaces sensitized by organic dyes with different anchoring groups: adsorption modes, electronic structure and implication for electron injection/recombination**  
(2012) *Phys. Chem. Chem. Phys.*, 14 (2), pp. 920-928.
- Xu, Y., Schoonen, A.A.  
**The absolute positions of conduction and valence bands of selected semiconductor minerals**  
(2000) *Am. Miner.*, 85, pp. 543-556.
- Skompska, M., Zarębska, K.  
**Electrodeposition of ZnO nanorod arrays on transparent conducting substrates – a review**  
(2014) *Electrochim. Acta*, 127, pp. 467-488.
- Znaidi, L.  
**Sol-gel-deposited ZnO thin films: a review**  
(2010) *Mater. Sci. Eng. B Solid State Mater. Adv. Technol.*, 174 (1-3), pp. 18-30.
- Kanoun, M., Goumri-Said, S., Manchon, A., Schwingenschlögl, U.  
**Ferromagnetism carried by highly delocalized hybrid states in Sc-doped ZnO thin films**  
(2012) *Appl. Phys. Lett.*, 100 (22), p. 222406.
- Xu, C., Wu, J., Desai, U.V., Gao, D.  
**Multilayer assembly of nanowire arrays for dye-sensitized solar cells**  
(2011) *J. Am. Chem. Soc.*, 133 (21), pp. 8122-8125.
- Payne, M.C., Teter, A., Arias, Joannopoulos  
**Iterative minimization techniques for ab initio total energy calculations: molecular dynamics and conjugate gradients**  
(1992) *Rev. Mod. Phys.*, 64 (4), pp. 1045-1097.
- De Angelis, F., Fantacci, S., Mosconi, E., Nazeeruddin, M.K., Grätzel, M.  
**Absorption spectra and excited state energy levels of the N719 dye on TiO<sub>2</sub> in dye-sensitized solar cell models**  
(2011) *J. Phys. Chem. C*, 115 (17), pp. 8825-8831.
- Delley, B.  
(1990) *J. Chem. Phys.*, 92.

B. Dellay, *J. Chem. Phys.*, 113 (2000), p. 7756

- Chaitanya, K., Heron, B.M., Ju, X.H.  
(2017) **Influence of a Local Electric Field on the Light Harvesting Efficiency of a Cyclopentadithiophene-Bridged D-A- $\pi$ -A Indoline Dye on Pure and N-Doped TiO<sub>2</sub> Surfaces**, 141.  
Elsevier Ltd
- Guo, M., Yang, H., Jian, X., Li, J., Liang, Z., Han, P.  
**The adsorptions of fixed groups -CN, -NH<sub>2</sub>, -SH, -OH and -COOH of dye molecules on stoichiometric, oxygen vacancy and Pt-doped SnO<sub>2</sub> (110) surfaces**  
(2018) *Appl. Surf. Sci.*, 428 (110), pp. 851-860.
- Perdew, J.P., Wang, Y.  
**Accurate and simple analytic representation of the electron-gas correlation energy**  
(1992), 45 (23), pp. 244-249.
- Cossi, M., Barone, V.  
**Time-dependent density functional theory for molecules in liquid solutions**  
(2001) *J. Chem. Phys.*, 115 (10), pp. 4708-4717.
- Studio, T.M.  
**The Materials Studio,”**  
(2001),
- Appalakondaiah, S., Vaitheeswaran, G., Lebègue, S., Christensen, N.E., Svane, A.  
**Effect of van der Waals interactions on the structural and elastic properties of black phosphorus**  
(2012) *Phys. Rev. B Condens. Matter Mater. Phys.*, 86 (3), pp. 1-9.
- Wang, P., Zakeeruddin, S.M., Moser, J.E., Nazeeruddin, M.K., Sekiguchi, T., Grätzel, M.  
**A stable quasi-solid-state dye-sensitized solar cell with an amphiphilic ruthenium sensitizer and polymer gel electrolyte**  
(2003) *Nat. Mater.*, 2 (6), pp. 402-407.
- Chen, C.-Y., Wu, S.-J., Wu, C.-G., Chen, J.-G., Ho, K.-C.  
**A ruthenium complex with superhigh light-harvesting capacity for dye-sensitized solar cells**  
(2006) *Angew. Chem.*, 118 (35), pp. 5954-5957.
- Suhaimi, S., Shahimin, M.M., Alahmed, Z.A., Chyský, J., Reshak, A.H.  
**Materials for enhanced dye-sensitized solar cell performance: electrochemical application**  
(2015) *Int. J. Electrochem. Sci.*, 10, pp. 2859-2871.
- Ihara, M., Tanaka, K., Sakaki, K., Honma, I., Yamada, K.  
**“Enhancement of the absorption coefficient of cis-(NCS)<sub>2</sub> bis(2,2'-bipyridyl-4,4'-dicarboxylate)ruthenium(II) dye in dye-sensitized solar cells by a Silver Island film**  
(1997) *J. Phys. Chem. B*, 101 (26), pp. 5153-5157.
- Han, W.G.  
**A theoretical study of the UV/visible absorption and emission solvatochromic properties of solvent-sensitive dyes**  
(2003) *ChemPhysChem*, 4 (10), pp. 1084-1094.

- Zhang, W., Wu, J., Wen, Y., Wu, W., Wang, L.  
(2018) *First Principles Study on Interface between Dual-Channel Anchorable Organic Dyes and TiO<sub>2</sub> for Dye-Sensitized Solar Cells*, 149.  
Elsevier Ltd
- Benghia, A., Dahame, T., Bentría, B.  
**First principle calculation of physical properties of barium based chalcogenides BaM<sub>4</sub>S<sub>7</sub> (M = Ga, Al); A DFT, DFT-D and hybrid functional HSE06 study**  
(2016) *Opt. Mater.*, 54, pp. 269-275.
- Serpone, N.  
**Is the band gap of pristine TiO<sub>2</sub> narrowed by anion- and cation-doping of titanium dioxide in second-generation photocatalysts?**  
(2006) *J. Phys. Chem. B*, 110 (48), pp. 24287-24293.
- Li, Z., Shi, F., Ding, Y., Zhang, T., Yan, C.  
**Facile Synthesis of Highly Ordered Mesoporous ZnTiO<sub>3</sub> with Crystalline Walls by Self-Adjusting Method**  
(2011), pp. 14589-14593.
- Ruiyu, J., Dengyang, L., Chao, C., Tao, Z., Fanfan, Y., Jinhong, Z.  
**Research article adsorption simulation of sulfur oxide on the surface of metal**  
(2014) *J. Chem. Pharm. Res.*, 6 (3), pp. 949-954.
- Yang, H.  
**First principles study of ruthenium sensitizer adsorption on anatase TiO<sub>2</sub> (001) surface**  
(2015) *RSC Adv.*, 5 (74), pp. 60230-60236.
- Azpiroz, J.M., De Angelis, F.  
**DFT/TDDFT study of the adsorption of N3 and N719 dyes on ZnO (1010) surfaces**  
(2014) *J. Phys. Chem. A*, 118 (31), pp. 5885-5893.
- Sun, C., Li, Y., Song, P., Ma, F.  
**An experimental and theoretical investigation of the electronic structures and photoelectrical properties of ethyl red and carminic acid for DSSC application**  
(2016) *Materials*, 9 (10), pp. 1-22.
- Hill, J.P.  
**Molecular Engineering Combined with Cosensitization Leads to Record Photovoltaic Efficiency for Non-ruthenium Solar Cells**  
(2016), *Angewandte Chemie – International Edition*
- Li, W.  
**What makes hydroxamate a promising anchoring group in dye-sensitized solar cells? Insights from theoretical investigation**  
(2014) *J. Phys. Chem. Lett.*, 5 (22), pp. 3992-3999.
- Yong, X., Schoonen, M.A.A.  
**The Absolute Energy Positions of Conduction and Valence Bands of Selected Semiconducting Minerals**  
(2000), " *Am. Mineral.*
- Zhu, H.C., Zhang, J., Wang, Y.L.  
**Adsorption orientation effects of porphyrin dyes on the performance of DSSC: comparison of benzoic acid and tropolone anchoring groups binding onto the TiO**

**<sub>2</sub> anatase (101) surface**

(2018) *Appl. Surf. Sci.*, 433 (101), pp. 1137-1147.

- Zhang, W., Wu, J., Wen, Y., Wu, W., Wang, L.  
**First Principles Study on Interface between Dual-Channel Anchorable Organic Dyes and TiO <sub>2</sub> for Dye-Sensitized Solar Cells**  
(2018), " *Dye. Pigment.*
- Sundari, C.D.D., Martoprawiro, M.A., Ivansyah, A.L.  
**A DFT and TDDFT study of PCM effect on N3 dye absorption in ethanol solution**  
(2017) *J. Phys. Conf. Ser.*, 812, p. 1.
- Fantacci, S., De Angelis, F., Selloni, A.  
**Absorption spectrum and solvatochromism of the [Ru(4,4'-COOH-2,2'-bpy)<sub>2</sub>(NCS)<sub>2</sub>] molecular dye by time dependent density functional theory**  
(2003) *J. Am. Chem. Soc.*, 2 (4), pp. 4381-4387.

**Publisher:** Elsevier Ltd

**ISSN:** 24686069

2-s2.0-85066014765

**Document Type:** Article

**Publication Stage:** Final

**Source:** Scopus

---

**ELSEVIER**

Copyright © 2019 Elsevier B.V. All rights reserved. Scopus® is a registered trademark of Elsevier B.V.

 RELX Group™